

## Synthesis and Transformations of 2-R-5-Aryl-5,6-dihydro-7H-[1,2,4]-triazolo[5,1-b]-[1,3]thiazin-7-ones

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**Abstract**—A new procedure for preparation of 2-R-5-aryl-5,6-dihydro-7H-[1,2,4]triazolo[5,1-b][1,3]thiazin-7-ones by condensation of 5-R-1,2,4-triazole-3-thiones with 3-arylacryloyl chlorides was developed. The thiazine ring of the [1,2,4]triazolo-[5,1-b][1,3]thiazin-7-ones is easily cleaved by treating with ammonia and hydrazine affording amides and hydrazides of 3-aryl-3-(1H-1,2,4-triazol-5-ylsulfanyl)propanoic acids. The latter react with isothiocyanates furnishing carbamoyl thiohydrazides of 3-aryl-3-(1H-1,2,4-triazol-5-ylsulfanyl)propanoic acids that in alkaline media undergo cyclization into 4-aryl-5-[2-(4H-1,2,4-triazol-5-ylsulfanyl)-2-phenylethyl]-2,4-dihydro-3H-1,2,4-triazole-5-thiones.

Derivatives of 4H-1,3-thiazin-4-one possess anti-phlogistic, analgesic, and tuberculocidal activity [1–3]. A 1,3-thiazine fragment is also included into the composition of 7-aminocephalosporanic acid whose structure underlies that of  $\beta$ -lactam antibiotics [4]. In fused heterocycles containing a 1,3-thiazine ring was recently revealed also antitumor [5] and anticonvulsant [6] activity. Therefore the synthesis and the study of properties of 1,3-thiazine derivatives is an urgent task.

1,2,4-Thiazole-3-thione is frequently used as an initial compound for the synthesis of 4H-1,3-thiazin-4-one derivatives [7–9]. It was shown formerly that the acylation of 1,2,4-triazoline-3-thione with acyl chlorides occurred as selective N-acylation into the position 2 [10, 11], and at the sacrifice of the mercapto group the 1,2,4-triazoline-3-thione readily added at 20°C to compounds containing an activated multiple bond, e.g., to propiolic and acrylic acids [9, 12], and to acrylonitrile [13]. We established [14] that the condensation of triazolinethione **Ia** with cinnamoyl chloride **IIa** gave rise to 5-aryl-5,6-dihydro-7H-[1,2,4]triazolo[5,1-b][1,3]-thiazin-7-one **IVa**. In continuation of this research we established that in this reaction were also involved 5-aryl-2,4-dihydro-3H-1,2,4-triazole-3-thiones **Ib–g**, 3-arylacryloyl chlorides **IIb, c**, and 3-heterylacryloyl chlorides **IId–f**. Thus were obtained 2-R-5-aryl-5,6-dihydro-7H-[1,2,4]-triazolo[5,1-b]-[1,3]-thiazin-7-ones **IVb–k** (Scheme 1).

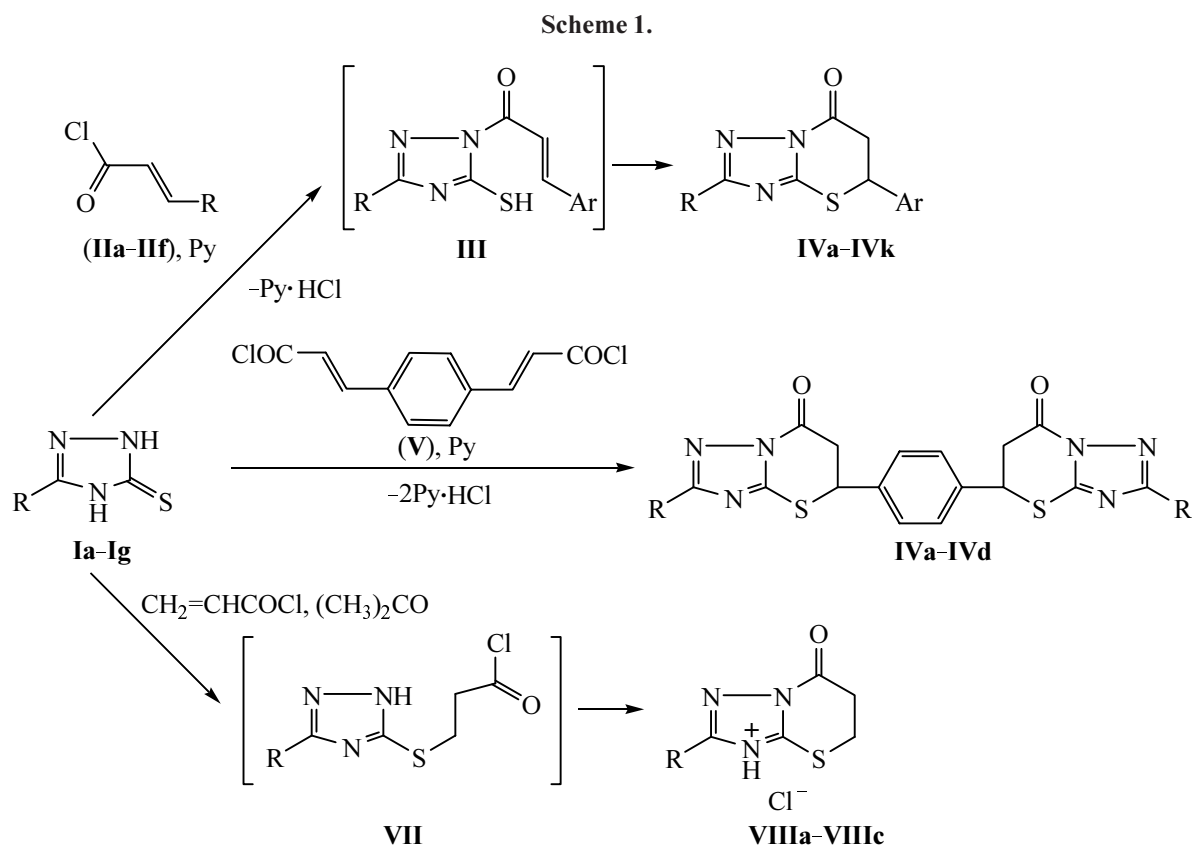
Triazolo[5,1-b][1,3]-thiazin-7-ones **IV** synthesized are colorless or yellowish substances well crystallizable from

acetic acid. Yields, melting points, elemental analyses, <sup>1</sup>H and IR spectra are compiled in Tables 1–3.

The heterocyclization occurs at heating at reflux a solution of initial compounds in a mixture benzene–pyridine, 1:1, for 1 h resulting in good yields (50–81%). The advantage of the procedure consists in the possibility to synthesize triazolo[5,1-b][1,3]thiazin-7-ones **IV** with various substituents (aryl, heteryl, and alkyl) in positions 2 and 5. The best yields of compounds **IV** were obtained at R = H, C<sub>6</sub>H<sub>5</sub>, Ar = C<sub>6</sub>H<sub>5</sub>, 4-CH<sub>3</sub>OC<sub>6</sub>H<sub>4</sub>, 3-O<sub>2</sub>NC<sub>6</sub>H<sub>4</sub> (65–81%), somewhat worse at R = alkyl, heteryl, Ar = heteryl (50–63%).

The data of <sup>1</sup>H NMR spectroscopy unambiguously prove the formation of a heterocycle. The signals of protons at the double bond of the initial 3-arylacryloyl chloride **II** in the <sup>1</sup>H NMR spectra are observed as two doublets in the region 6.80 and 7.50 ppm. The double C=C bond after heterocyclization becomes an ordinary bond in compounds **IV**. The signals of protons in positions 5 and 6 appear in the <sup>1</sup>H NMR spectra as ABX system, and their signals are present in the region 3.30–5.50 ppm. The characteristic absorption bands in the IR spectra of compounds **IV** are those of C=O (1720–1750 cm<sup>-1</sup>) and C=N (1590–1610 cm<sup>-1</sup>) groups.

Notice that the attempt to carry out the cyclization of 1,2,4-triazole-3-thiones **I** with 3-arylacryloyl chlorides **II** in acetone or benzene failed. The reaction does not occur at room temperature, and at heating only tarring of the reaction mixture was observed. Therefore it is presumable



**I**, R = H (**a**), C<sub>6</sub>H<sub>5</sub> (**b**), 4-FC<sub>6</sub>H<sub>4</sub> (**c**), 1-naphthyl (**d**), 1-adamantyl (**e**), (C<sub>6</sub>H<sub>5</sub>)<sub>2</sub>CH (**f**), 1,3-benzodioxol-5-yl (**g**); **II**, Ar = C<sub>6</sub>H<sub>5</sub> (**a**), 4-CH<sub>3</sub>OC<sub>6</sub>H<sub>4</sub> (**b**), 3-O<sub>2</sub>NC<sub>6</sub>H<sub>4</sub> (**c**), 1,3-benzodioxol-5-yl (**d**), 2-thienyl (**e**), 2-furyl (**f**); **IV**, R = H (**a-c**), C<sub>6</sub>H<sub>5</sub> (**d**), 4-FC<sub>6</sub>H<sub>4</sub> (**e, f**), 1-naphthyl (**g**), 1-adamantyl (**h, i**), (C<sub>6</sub>H<sub>5</sub>)<sub>2</sub>CH (**j**), 1,3-benzodioxol-5-yl (**k**); Ar = C<sub>6</sub>H<sub>5</sub> (**a, e, g, h, j, k**), 4-CH<sub>3</sub>OC<sub>6</sub>H<sub>4</sub> (**b**), 3-O<sub>2</sub>NC<sub>6</sub>H<sub>4</sub> (**c**), 1,3-benzodioxol-5-yl (**d**), 2-thienyl (**f**), 2-furyl (**i**); **VI**, R = C<sub>6</sub>H<sub>5</sub> (**a**), 4-CH<sub>3</sub>OC<sub>6</sub>H<sub>4</sub> (**b**), 4-CH<sub>3</sub>C<sub>6</sub>H<sub>4</sub> (**c**), 4-FC<sub>6</sub>H<sub>4</sub> (**d**); **VIII**, R = C<sub>6</sub>H<sub>5</sub> (**a**), 4-CH<sub>3</sub>OC<sub>6</sub>H<sub>4</sub> (**b**), 4-FC<sub>6</sub>H<sub>4</sub> (**c**).

[15] that the reaction of 1,2,4-triazole-3-thiones **I** with 3-arylacryloyl chlorides in pyridine proceeds first as the N-acylation of triazolone-3-thione to give intermediate adduct **III** that then converts into triazolo[5,1-*b*]-[1,3]-thiazin-7-one **IV**.

In the same fashion 1,2,4-triazole-3-thiones **I** react with 3-{4-[2-(chlorocarbonyl)ethenyl]phenyl}acryloyl chloride **V** in pyridine. As a result of this condensation form benzene-1,4-diylbis-2-R-5-aryl-5,6-dihydro-7H-[1,2,4]triazolo[5,1-*b*]-[1,3]-thiazin-7-ones **IVa-d** in 61–72% yield.

Yet 1,2,4-triazole-3-thiones **I** do not react with acryloyl-chloride in pyridine. The reaction of 1,2,4-triazole-3-thiones **I** with acryloyl-chloride takes place in acetone giving rise to triazolo[5,1-*b*][1,3]-thiazin-7-ones hydrochlorides **VIII** (yield 60–71%). This reaction apparently occurs [16] first as an addition of 1,2,4-triazolone-3-thione by its thio group to the double bond

of the acryloyl-chloride via intermediate **VII**, and then it undergoes an intramolecular acylation affording compound **VIII**.

In the <sup>1</sup>H NMR spectra of compounds **VIIIa-c** as characteristic signals should be mentioned two triplets from the protons of the thiazine ring (δ 2.82–2.87 and 3.33–3.38 ppm), and the broadened singlet from the proton of the NH group (δ 6.80–7.40 ppm).

Different conditions required by reactions of 5-aryl-1,2,4-triazole-3-thione with 3-arylacryloyl chlorides (in pyridine) and with acryloyl-chloride (in acetone) although both processes afford [1,2,4]triazolo[5,1-*b*][1,3]thiazines are apparently due to unequal reactivity of the double bonds in 3-arylacryloylchlorides and acryloylchloride.

In the same fashion 1,2,4-triazole-3-thiones **I** react with 3-{4-[2-(chlorocarbonyl)ethenyl]phenyl}acryloyl chloride **V** in pyridine. As a result of this condensation form benzene-1,4-diylbis-2-R-5-aryl-5,6-dihydro-7H-

**Table 1.** Yields, melting points, and elemental analyses of 5,6-dihydro-7*H*-[1,2,4]triazolo[5,1-*b*][1,3]thiazin-7-ones **IVa–k**, **VIa–d**, and **VIIIa–c**, propanamides **IXa, b**, propanehydrazides **Xa, b**, hydrazones **XIa, b**, thiosemicarbazides **XIIa, b**, and 1,2,4-triazole-3-thione **XIII**

Compd. no.	Yield, %	mp, °C <sup>a</sup>	Found, %			Formula	Calculated, %		
			C	H	N		C	H	N
<b>IVa</b>	70	161–163	56.86	3.95	18.20	C <sub>11</sub> H <sub>9</sub> N <sub>3</sub> OS	57.14	3.92	18.18
<b>IVb</b>	72	143–145	55.12	4.10	16.12	C <sub>12</sub> H <sub>11</sub> N <sub>3</sub> O <sub>2</sub> S	55.17	4.24	16.09
<b>IVc</b>	81	202–205	47.85	2.90	20.60	C <sub>11</sub> H <sub>8</sub> N <sub>4</sub> O <sub>3</sub> S	47.83	2.92	20.29
<b>IVd</b>	70	190–192	61.77	3.61	12.12	C <sub>18</sub> H <sub>13</sub> N <sub>3</sub> O <sub>3</sub> S	61.53	3.73	11.96
<b>IVe</b>	68	193–195	62.55	3.86	13.05	C <sub>17</sub> H <sub>12</sub> FN <sub>3</sub> OS	62.76	3.72	12.91
<b>IVf</b>	61	188–190	54.14	2.91	12.80	C <sub>15</sub> H <sub>10</sub> FN <sub>3</sub> OS <sub>2</sub>	54.37	3.04	12.68
<b>IVg</b>	57	137–139	70.41	4.52	11.89	C <sub>21</sub> H <sub>15</sub> N <sub>3</sub> OS	70.57	4.23	11.76
<b>IVh</b>	69	220–222	69.30	6.21	11.69	C <sub>21</sub> H <sub>23</sub> N <sub>3</sub> OS	69.01	6.34	11.50
<b>IVi</b>	67	217–220	64.09	6.13	12.02	C <sub>19</sub> H <sub>21</sub> N <sub>3</sub> O <sub>2</sub> S	64.20	5.95	11.82
<b>IVj</b>	50	163–165	72.38	4.98	10.71	C <sub>24</sub> H <sub>19</sub> N <sub>3</sub> OS	72.52	4.82	10.57
<b>IVk</b>	52	190–192	61.75	3.89	12.15	C <sub>18</sub> H <sub>13</sub> N <sub>3</sub> O <sub>3</sub> S	61.53	3.73	11.96
<b>VIa</b>	54	277–280	62.84	3.97	15.44	C <sub>28</sub> H <sub>20</sub> N <sub>6</sub> O <sub>2</sub> S <sub>2</sub>	62.67	3.76	15.66
<b>VIb</b>	68	330–335	60.04	4.19	14.22	C <sub>30</sub> H <sub>24</sub> N <sub>6</sub> O <sub>4</sub> S <sub>2</sub>	60.39	4.05	14.08
<b>VIc</b>	57	316–320	63.97	4.41	14.70	C <sub>30</sub> H <sub>24</sub> N <sub>6</sub> O <sub>2</sub> S <sub>2</sub>	63.81	4.28	14.88
<b>VId</b>	51	310–315	58.65	3.33	14.78	C <sub>28</sub> H <sub>18</sub> F <sub>2</sub> N <sub>6</sub> O <sub>2</sub> S <sub>2</sub>	58.73	3.17	14.68
<b>VIIIa</b>	67	137–140	49.48	3.68	15.77	C <sub>11</sub> H <sub>10</sub> ClN <sub>3</sub> OS	49.35	3.76	15.69
<b>VIIIb</b>	55	141–143	48.62	4.17	14.25	C <sub>12</sub> H <sub>12</sub> ClN <sub>3</sub> O <sub>2</sub> S	48.41	4.06	14.11
<b>VIIIc</b>	61	120–122	46.09	3.34	14.83	C <sub>11</sub> H <sub>9</sub> ClFN <sub>3</sub> OS	46.24	3.17	14.71
<b>IXa</b>	77	143–145	53.42	5.08	22.37	C <sub>11</sub> H <sub>12</sub> N <sub>4</sub> OS	53.21	4.87	22.56
<b>IXb</b>	71	153–155	51.60	5.29	20.29	C <sub>12</sub> H <sub>14</sub> N <sub>4</sub> O <sub>2</sub> S	51.79	5.07	20.13
<b>Xa</b>	83	163–165	50.32	5.13	26.78	C <sub>11</sub> H <sub>13</sub> N <sub>5</sub> OS	50.18	4.98	26.60
<b>Xb</b>	75	170–172	49.27	5.28	24.06	C <sub>12</sub> H <sub>15</sub> N <sub>5</sub> O <sub>2</sub> S	49.13	5.15	23.87
<b>XIa</b>	68	204–207	61.57	5.06	20.14	C <sub>18</sub> H <sub>17</sub> N <sub>5</sub> OS	61.52	4.88	19.93
<b>XIb</b>	59	185–187	53.61	4.10	19.78	C <sub>16</sub> H <sub>15</sub> N <sub>5</sub> OS <sub>2</sub>	53.76	4.23	19.59
<b>XIIa</b>	83	190–194	54.29	4.31	21.25	C <sub>18</sub> H <sub>18</sub> N <sub>6</sub> OS <sub>2</sub>	54.25	4.55	21.09
<b>XIIb</b>	75	187–190	46.30	4.93	25.20	C <sub>15</sub> H <sub>16</sub> N <sub>6</sub> OS <sub>2</sub>	46.41	4.79	24.98
<b>XIII</b>	63	145–147	56.87	4.13	21.91	C <sub>18</sub> H <sub>16</sub> N <sub>6</sub> S <sub>2</sub>	56.82	4.24	22.09

<sup>a</sup> Compounds **IVa–IVk**, **XIa**, **XIb**, and **XIII** were recrystallized from acetic acid, **VIa–VId** from benzonitrile, **VIIIa–VIIIc** from ethanol, **IXa**, **IXb** and **Xa, b** from a mixture ethanol–water, 1 : 1.

[1,2,4]triazolo[5,1-*b*][1,3]thiazin-7-ones **VIa–VId** in 61–72% yield.

Triazolo[5,1-*b*][1,3]-thiazin-7-ones are poorly studied compounds whose chemical properties are hardly described. We found that the CO–N bond of the 1,3-thiazine ring was weak, and triazolo[5,1-*b*][1,3]-thiazin-7-ones **IV** readily reacted with ammonia and hydrazine. After these reactions we isolated from the reaction mixtures 3-aryl-3-(1*H*-1,2,4-triazol-5-ylsulfanyl)propanoic acids amides **IXa**, **IXb** and hydrazides **Xa**, **Xb**.

Hydrazides **X** possess high reactivity and form hydrazones **XIa**, **XIb** with aldehydes and with isothiocyanates afford thiosemicarbazide derivatives **XIIa**, **XIIb**. In a boiling water solution of sodium hydroxide compound **XIIa** converted into 5-[2-(1*H*-1,2,4-triazol-5-ylsulfanyl)-2-phenylethyl]-4-phenyl-2,4-dihydro-3*H*-1,2,4-triazole-3-thione **XIII** (Scheme 2).

In the <sup>1</sup>H NMR spectra of compounds **IXa**, **IXb** and **Xa**, **Xb** the following signals are characteristic: peaks of CH<sub>2</sub>–CH groups (*ABX* system, δ 2.76–2.90 and 4.97–

**Table 2.** <sup>1</sup>H NMR spectra of 5,6-dihydro-7H-[1,2,4]triazolo[5,1-b][1,3]thiazin-7-ones **IVa–IVk**, **VIa–VIId**, and **VIIIa–VIIIc**, propanamides **IXa**, **IXb**, propanehydrazides **Xa**, **Xb**, hydrazones **XIa**, **XIb**, thiosemicarbazides **XIIa**, **XIIb**, and 1,2,4-triazole-3-thione **XIII**

Compd. no.	<sup>1</sup> H NMR spectrum (DMSO- <i>d</i> <sub>6</sub> ), δ, ppm ( <i>J</i> , Hz) <sup>a</sup>
<b>IVa</b>	3.42 m (1H, H <sup>6</sup> ), 3.90 m (1H, H <sup>6</sup> ), 5.45 (1H, H <sup>5</sup> ), 7.40–7.62 m (5H, C <sub>6</sub> H <sub>5</sub> ), 8.31 s (1H, H <sup>2</sup> )
<b>IVb</b>	3.41 m (1H, H <sup>6</sup> ), 3.78 m (1H, H <sup>6</sup> ), 3.83 c (3H, CH <sub>3</sub> O), 5.36 m (1H, H <sup>5</sup> ), 7.00 d (2H, H arom, <i>J</i> 8.1), 7.53 d (2H, H arom, <i>J</i> 8.1), 8.06 s (1H, H <sup>2</sup> )
<b>IVc</b>	3.41 m (1H, H <sup>6</sup> ), 3.99 m (1H, H <sup>6</sup> ), 5.60 m (1H, H <sup>5</sup> ), 7.76 t (1H, H arom, <i>J</i> 8.7), 7.95 d (1H, H arom, <i>J</i> 8.7), 8.25 d (1H, H arom, <i>J</i> 8.7) 8.30 c (1H, H <sup>2</sup> ), 8.38 d (1H, H arom, <i>J</i> 8.7)
<b>IVd</b>	3.38 m (1H, H <sup>6</sup> ), 3.91 m (1H, H <sup>6</sup> ), 5.41 m (1H, H <sup>5</sup> ), 6.07 s (2H, OCH <sub>2</sub> O), 6.98 m (2H, H arom), 7.13 s (1H, H arom), 7.55 m (3H, C <sub>6</sub> H <sub>5</sub> ), 8.09 m (2H, C <sub>6</sub> H <sub>5</sub> )
<b>IVe</b>	3.46 m (1H, H <sup>6</sup> ), 3.89 m (1H, H <sup>6</sup> ), 5.51 m (1H, H <sup>5</sup> ), 7.31–7.59 m (7H, H arom), 8.13 m (2H, H arom)
<b>IVf</b>	3.66 m (1H, H <sup>6</sup> ), 3.83 m (1H, H <sup>6</sup> ), 5.76 m (1H, H <sup>5</sup> ), 7.04 d.d (1H, H <sup>1</sup> thiophene, <i>J</i> <sub>1</sub> 4.8, <i>J</i> <sub>2</sub> 3.1), 7.21 d (1H, H <sup>3</sup> thiophene, <i>J</i> 3.1), 7.40 m (2H, H arom), 7.57 d (1H, H <sup>2</sup> thiophene, <i>J</i> 4.8), 8.11 m (2H, H arom)
<b>IVg</b>	3.49 m (1H, H <sup>6</sup> ), 3.99 m (1H, H <sup>6</sup> ), 5.55 m (1H, H <sup>5</sup> ), 7.42–7.70 m (8H, H arom), 8.06 d (1H, C <sub>10</sub> H <sub>7</sub> , <i>J</i> 9.0), 8.15 d (1H, C <sub>10</sub> H <sub>7</sub> , <i>J</i> 9.0), 8.30 d (1H, C <sub>10</sub> H <sub>7</sub> , <i>J</i> 9.0), 9.14 d (1H, C <sub>10</sub> H <sub>7</sub> , <i>J</i> 9.0)
<b>IVh</b>	1.73 s (6H, C <sub>10</sub> H <sub>15</sub> ), 1.93 s (6H, C <sub>10</sub> H <sub>15</sub> ), 2.03 s (3H, C <sub>10</sub> H <sub>15</sub> ), 3.22 m (1H, H <sup>6</sup> ), 3.83 m (1H, H <sup>6</sup> ), 5.40 m (1H, H <sup>5</sup> ), 7.42 m (5H, C <sub>6</sub> H <sub>5</sub> )
<b>IVi</b>	1.72 s (6H, C <sub>10</sub> H <sub>15</sub> ), 1.92 s (6H, C <sub>10</sub> H <sub>15</sub> ), 2.02 s (3H, C <sub>10</sub> H <sub>15</sub> ), 3.58 d (2H, H <sup>6</sup> , <i>J</i> 6.0), 5.44 t (1H, H <sup>5</sup> , <i>J</i> 6.0), 6.43 m (2H, furan), 7.69 d (1H, H <sup>3</sup> furan, <i>J</i> 2.9)
<b>IVj</b>	3.29 m (1H, H <sup>6</sup> ), 3.87 m (1H, H <sup>6</sup> ), 5.47 m (1H, H <sup>5</sup> ), 5.68 s [1H, (C <sub>6</sub> H <sub>5</sub> ) <sub>2</sub> CH], 7.03–7.69 m (15H, 3C <sub>6</sub> H <sub>5</sub> )
<b>IVk</b>	3.40 m (1H, H <sup>6</sup> ), 3.85 m (1H, H <sup>6</sup> ), 5.51 m (1H, H <sup>5</sup> ), 6.13 s (2H, OCH <sub>2</sub> O), 7.05 d (1H, H arom, <i>J</i> 7.6), 7.40–7.65 m (7H, H arom)
<b>VIa</b>	2.98–3.68 m (4H, 2CH <sub>2</sub> CO), 5.13 m (2H, 2SCH), 7.03 s (4H, H arom), 7.32–7.71 m (10H, 2C <sub>6</sub> H <sub>5</sub> )
<b>VIb</b>	3.20–3.61 m (4H, 2CH <sub>2</sub> CO), 3.79 s (6H, 2CH <sub>3</sub> O), 5.39 m (2H, 2SCH), 6.96 d (4H, H arom, <i>J</i> 8.0), 7.52 d (4H, H arom, <i>J</i> 8.1), 7.79 d (4H, H arom, <i>J</i> 8.1)
<b>VIc</b>	2.24 s (6H, 2CH <sub>3</sub> ), 3.42–3.83 m (4H, 2CH <sub>2</sub> CO), 5.32 m (2H, 2SCH), 7.18 d (4H, H arom, <i>J</i> 6.9), 7.46 s (4H, H arom), 7.61 d (4H, H arom, <i>J</i> 6.9)
<b>VIId</b>	3.45–3.89 m (4H, 2CH <sub>2</sub> CO), 5.35 m (2H, 2SCH), 7.05 s (4H, H arom), 7.49 s (4H, H arom), 7.81 s (4H, H arom)
<b>VIIIa</b>	2.85 t (2H, CH <sub>2</sub> CO, <i>J</i> 6.5), 3.37 t (2H, CH <sub>2</sub> S, <i>J</i> 6.5), 6.03 br.s (1H, HCl), 7.51 m (3H, C <sub>6</sub> H <sub>5</sub> ), 8.02 m (2H, C <sub>6</sub> H <sub>5</sub> )
<b>VIIIb</b>	2.83 t (2H, CH <sub>2</sub> CO, <i>J</i> 6.2), 3.34 t (2H, CH <sub>2</sub> S, <i>J</i> 6.2), 3.81 s (3H, CH <sub>3</sub> O), 6.80 br.s (1H, HCl), 7.12 d (2H, H arom, <i>J</i> 8.5), 8.02 d (2H, H arom, <i>J</i> 8.5)
<b>VIIIc</b>	2.87 t (2H, CH <sub>2</sub> CO, <i>J</i> 6.7), 3.37 t (2H, CH <sub>2</sub> S, <i>J</i> 6.7), 7.01 br.s. (1H, HCl), 7.40 m (2H, H arom), 8.09 m (2H, H arom)
<b>IXa</b>	2.81 d (2H, CH <sub>2</sub> CO, <i>J</i> 5.5), 5.01 t (1H, SCH, <i>J</i> 5.5), 6.92 br.s (1H, CONH), 7.20 m (5H, C <sub>6</sub> H <sub>5</sub> ), 7.33 br.s (1H, CONH), 8.53 s (1H, CH triazole), 14.07 br.s (1H, NH triazole)
<b>IXb</b>	2.78 d (2H, CH <sub>2</sub> CO, <i>J</i> 6.2), 3.75 s (3H, CH <sub>3</sub> O), 4.98 t (1H, SCH, <i>J</i> 6.2), 6.98 br.s (1H, CO–NH), 6.79 d (2H, H arom, <i>J</i> 8.5), 7.25 d (2H, H arom, <i>J</i> 8.5), 7.31 br.s (1H, CO–NH), 8.38 s (1H, CH triazole), 13.91 br.s (1H, NH triazole)
<b>Xa</b>	2.81 d (2H, CH <sub>2</sub> CO, <i>J</i> 6.7), 3.80 br.s (2H, NH <sub>2</sub> ), 5.11 t (1H, SCH, <i>J</i> 6.7), 7.26 m (5H, C <sub>6</sub> H <sub>5</sub> ), 8.42 s (1H, CH triazole), 9.06 br.s (1H, CONH), 13.91 br.s (1H, NH triazole)
<b>Xb</b>	2.76 d (2H, CH <sub>2</sub> CO, <i>J</i> 6.6), 3.65 br.s (2H, NH <sub>2</sub> ), 3.78 s (3H, CH <sub>3</sub> O), 5.01 t (1H, SCH, <i>J</i> 6.6), 6.81 d (2H, H arom, <i>J</i> 7.7), 7.26 d (2H, H arom, <i>J</i> 7.7), 8.40 s (1H, CH triazole), 9.03 br.s (1H, CONH), 13.52 br.s (1H, NH triazole)
<b>XIa<sup>b</sup></b>	3.10 m (1H, CH <sub>2</sub> CO), 3.53 m (1H, CH <sub>2</sub> CO), 5.24 m (1H, SCH), 7.10–7.82 m (10H, 2C <sub>6</sub> H <sub>5</sub> ), 8.07 s and 8.24 s (1H, CH=N), 8.60 s (1H, CH triazole), 11.32 s and 11.51 s (1H, CONH), 14.14 br.s (1H, NH triazole)
<b>XIb<sup>b</sup></b>	3.26 m (1H, CH <sub>2</sub> CO), 3.51 m (1H, CH <sub>2</sub> CO), 5.13 m (1H, SCH), 7.05–7.76 m (8H, C <sub>6</sub> H <sub>5</sub> + C <sub>4</sub> H <sub>3</sub> S), 8.22 s and 8.32 s (1H, CH=N), 8.59 s (1H, CH triazole), 11.28 s and 11.33 s (1H, CONH), 14.12 br.s (1H, NH triazole)
<b>XIIa</b>	3.05 m (2H, CH <sub>2</sub> CO), 5.12 m (1H, SCH), 7.03–7.96 m (10H, 2C <sub>6</sub> H <sub>5</sub> ), 8.57 s (1H, CH triazole), 9.41 br.s (1H, C <sub>6</sub> H <sub>5</sub> NHCS), 9.59 br.s (1H, NHNHCS), 10.04 br.s (1H, CONH), 14.03 br.s (1H, NH triazole)
<b>XIIb</b>	2.71 d (3H, CH <sub>3</sub> , <i>J</i> 3.8), 3.02 m (2H, CH <sub>2</sub> CO), 5.06 m (1H, SCH), 7.05–7.60 m (6H, C <sub>6</sub> H <sub>5</sub> + NHCH <sub>3</sub> ), 8.59 s (1H, CH triazole), 9.23 br.s (1H, NHNHCS), 9.82 br.s (1H, CONH), 14.13 br.s (1H, NH triazole)
<b>XIII</b>	3.21 m (2H, CH <sub>2</sub> CO), 4.77 m (1H, SCH), 7.04–7.80 m (11H, 2C <sub>6</sub> H <sub>5</sub> + NHC=S), 8.42 s (1H, CH triazole), 13.71 br.s (1H, NH triazole)

<sup>a</sup> Spectrum of compound **IVb** was registered in acetone-*d*<sub>6</sub>, spectra of compounds **VIa–VIId** were recorded in CF<sub>3</sub>COOD.

<sup>b</sup> Hydrazones **XIa**, **XIb** form as a mixture of (*Z*)- and (*E*)-form in a ratio approximately 1:1.

**Table 3.** IR spectra of 5,6-dihydro-7*H*-[1,2,4]triazolo[5,1-*b*]-[1,3]thiazin-7-ones **IVa–IVk**, **VIa–VIId**, and **VIIIa–VIIIc**, propanamides **IXa**, **IXb**, propanehydrazides **Xa**, **Xb**, hydrazones **XIa**, **XIb**, thiosemicarbazides **XIIa**, **XIIb**, and 1,2,4-triazole-3-thione **XIII**

Compd. no.	IR spectrum, $\nu$ , $\text{cm}^{-1}$
<b>IVa</b>	3100–3000, 1730, 1600, 1510, 1450, 1410
<b>IVb</b>	3100–3000, 1720, 1600, 1480, 1420
<b>IVc</b>	3100–3000, 1730, 1590, 1540, 1510, 1400
<b>IVd</b>	3100–2900, 1740, 1610, 1480, 1440
<b>IVe</b>	3100–2900, 1750, 1600, 1490, 1410
<b>IVf</b>	3150–3000, 1740, 1600, 1480, 1420
<b>IVg</b>	3100–3000, 1740, 1610, 1500, 1480, 1450
<b>IVh</b>	3000–2800, 1730, 1600, 1500, 1450
<b>IVi</b>	3000–2800, 1720, 1620, 1500, 1450, 1410
<b>IVj</b>	3100–2900, 1730, 1590, 1520, 1490, 1450
<b>IVk</b>	3100–3000, 1720, 1600, 1490, 1440
<b>VIa</b>	3100–3000, 1740, 1600, 1590, 1470, 1410
<b>VIb</b>	3100–3000, 1750, 1600, 1580, 1480, 1420
<b>VIc</b>	3100–3000, 1730, 1610, 1530, 1490, 1410
<b>VIId</b>	3150–3000, 1750, 1600, 1490, 1410
<b>VIIIa</b>	3250–2700, 1730, 1610, 1580, 1500, 1440, 1410
<b>VIIIb</b>	3250–2700, 1720, 1610, 1590, 1520, 1480, 1430
<b>VIIIc</b>	3250–2700, 1710, 1610, 1510, 1490, 1410
<b>IXa</b>	3200–2800, 1660, 1610, 1490, 1440, 1400
<b>IXb</b>	3200–2800, 1650, 1600, 1510, 1440, 1410
<b>Xa</b>	3200–2800, 1630, 1590, 1550, 1480, 1450, 1430
<b>Xb</b>	3200–2800, 1680, 1610, 1520, 1460
<b>XIa</b>	3200–2800, 1660, 1600, 1490, 1450, 1400
<b>XIb</b>	3200–2800, 1670, 1590, 1550, 1490, 1430, 1400
<b>XIIa</b>	3200–2800, 1680, 1540, 1500, 1450
<b>XIIb</b>	3200–2800, 1670, 1560, 1510, 1440
<b>XIII</b>	3100–2900, 2500, 1600, 1510

5.11 ppm), of CO–NH<sub>2</sub> groups ( $\delta$  6.92–6.98 and 7.31–7.33 ppm), of CO–NH–NH<sub>2</sub> groups ( $\delta$  3.65–3.80 and 9.03–9.06 ppm), and of CH and NH protons of the triazole ring ( $\delta$  8.38–8.53 and 14.12–14.23 ppm). In the IR spectra appear characteristic absorption bands of NH groups (2800–3200  $\text{cm}^{-1}$ ), C=O groups (1630–1680  $\text{cm}^{-1}$ ), and C=N bonds (1590–1610  $\text{cm}^{-1}$ ). In the <sup>1</sup>H NMR spectra of compounds **XIa**, **XIb** and **XIIa**, **XIIb** are present signals from protons of CH<sub>2</sub>–CH groups (*ABX* system,  $\delta$  3.02–3.53 and 5.06–5.24 ppm), of CH=N group ( $\delta$  8.07–8.32 ppm), of CO–NH ( $\delta$  9.82–11.51 ppm), of CH and NH protons of triazole ring ( $\delta$  8.57–8.60 ppm and 14.03–14.14 ppm), NH–CS ( $\delta$  9.41 ppm), and NH–

NH–CS group ( $\delta$  9.23–9.59 ppm). The <sup>1</sup>H NMR spectrum of compound **XIII** is simpler than the spectra of initial compounds, and it contains signals of protons from CH<sub>2</sub>–CH group ( $\delta$  3.21–4.77 ppm), from aromatic rings ( $\delta$  7.04–7.80 ppm), and from protons of CH and NH groups of the triazole ring ( $\delta$  8.42 and 13.71 ppm).

And so the reaction of 5-R-1,2,4-triazole-3-thiones with acryloyl chloride, 3-arylacryloyl chlorides, and benzene-1,4-diylbisacryloyl chloride possesses a general character and is a new convenient one-stage synthetic procedure for preparation of [1,2,4]triazolo[5,1-*b*]-[1,3]thiazin-7-ones interesting as potential physiologically active substances. Being highly reactive they are promising for preparation of other heterocyclic compounds.

## EXPERIMENTAL

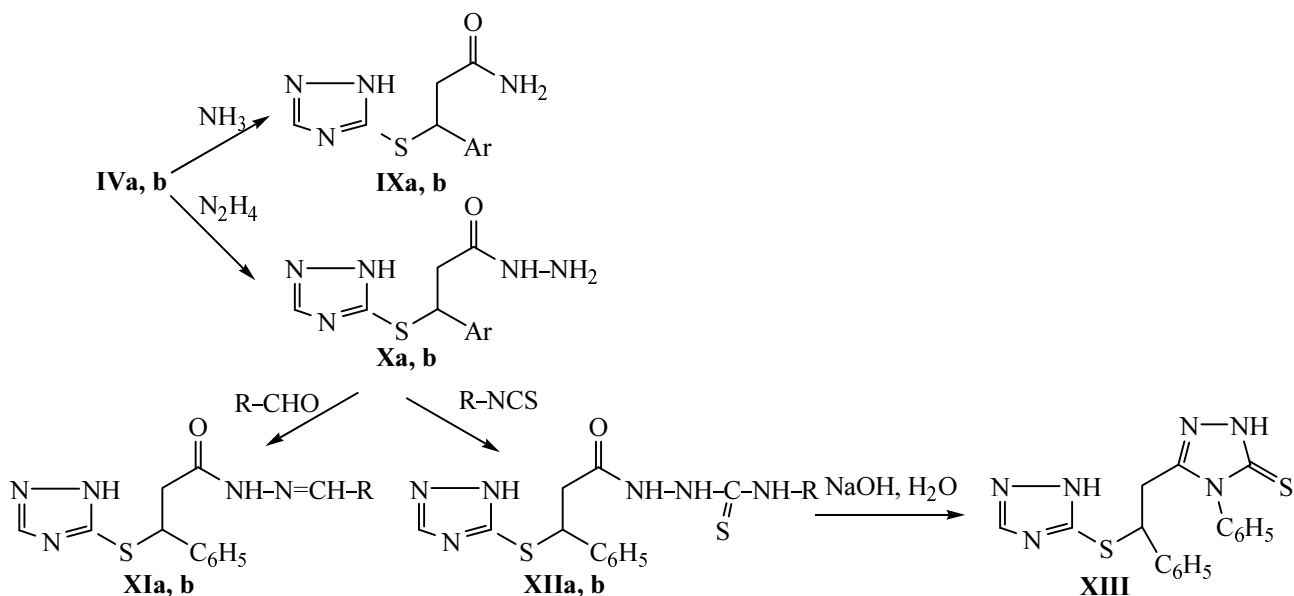
<sup>1</sup>H NMR spectra were registered on spectrometer Varian VXR-300 (operating frequency 300 MHz) from solutions of compounds in DMSO-*d*<sub>6</sub>, acetone-*d*<sub>6</sub>, or CF<sub>3</sub>COOD, internal reference TMS. IR spectra from samples of compounds pelletized with KBr were recorded on spectrophotometer UR-20.

**2-R-5-Aryl-5,6-dihydro-7*H*-[1,2,4]triazolo-[5,1-*b*]-[1,3]thiazin-7-ones (IVa–k), and benzene-1,4-diylbis(2-R-5-aryl-5,6-dihydro-7*H*-[1,2,4]triazolo[5,1-*b*][1,3]thiazin-7-ones (VIa–d).** To a solution of 10 mmol of 5-R-2,4-dihydro-3*H*-1,2,4-triazole-3-thione **I** in 5 ml of pyridine at 20°C was added a solution of 10 mmol of 3-arylacryloyl-chloride **II** or 5 mmol of benzene-1,4-diylbisacryloyl chloride **V** in 5 ml of benzene. The solution was heated at reflux for 1 h, cooled, and diluted with 50 ml of water. The separated precipitate was filtered off, washed with ether, and dried.

**2-R-5,6-Dihydro-7*H*-[1,2,4]triazolo[5,1-*b*][1,3]thiazin-7-ones hydrochlorides (VIIIa–c).** To a dispersion of 10 mmol of 5-R-2,4-dihydro-3*H*-1,2,4-triazole-3-thione **I** in 30 ml of acetone at 25°C was added 11 mmol of acryloyl-chloride, and the mixture was stirred for 4 h. The separated precipitate of compound **VIII** was filtered off, washed with acetone, and dried.

**3-Aryl-3-(1*H*-1,2,4-triazol-5-ylsulfanyl)-propanamides (IXa, b), and 3-aryl-3-(1*H*-1,2,4-triazol-5-ylsulfanyl)propanehydrazides (Xa, b).** Into a solution of 5 mmol of triazolo[5,1-*b*][1,3]thiazin-7-one **IV** in 10 ml of ethanol at 25°C for 1 h was passed a flow of 30 mmol of gaseous ammonia or was added dropwise

Scheme 2.



IV, IX, X, Ar = C<sub>6</sub>H<sub>5</sub> (a), 4-CH<sub>3</sub>OC<sub>6</sub>H<sub>4</sub> (b); XI, R = C<sub>6</sub>H<sub>5</sub> (a), 2-thienyl (b); XII, R = C<sub>6</sub>H<sub>5</sub> (a), CH<sub>3</sub> (b).

10 mmol of hydrazine hydrate. The reaction mixture was left standing for 24 h at room temperature, then it was heated for 3 h at 60°C. The reaction product **IX** or **X** precipitated on cooling was filtered off and dried.

**N<sup>1</sup>-(R-Methylene)-3-(1H-1,2,4-triazol-5-ylsulfanyl)-3-phenylpropanehydrazides (XIa, b).** To a solution of 2.5 mmol of hydrazide **X** in 4 ml of acetic acid was added at 40°C 2.6 mmol of an appropriate aldehyde, and the reaction mixture was left standing for 24 h at the above temperature. On cooling the separated precipitate of compound **XI** was filtered off, washed with ethanol, and dried.

**4R-1-[3-(1H-1,2,4-triazol-5-ylsulfanyl)-3-phenylpropanoyl]thiosemicarbazides (XIIa, b).** To a solution of 2.5 mmol of hydrazide **X** in 4 ml of ethanol at 20°C was added 2.6 mmol of an appropriate isothiocyanate. The reaction mixture was boiled for 2 h, then it was cooled, the separated precipitate of compound **XII** was filtered off and dried.

**5-[2-(1H-1,2,4-triazol-5-ylsulfanyl)-2-phenylethyl]-4-phenyl-2,4-dihydro-3H-1,2,4-triazole-3-thione (XIII).** A solution of 1.5 mmol of thiosemicarbazide XIIa and 3 mmol of sodium hydroxide in 5 ml of water was boiled for 2 h, then cooled, and acidified with 0.6 ml of 30% hydrochloric acid. The separated precipitate of compound **XIII** was filtered off and dried.

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